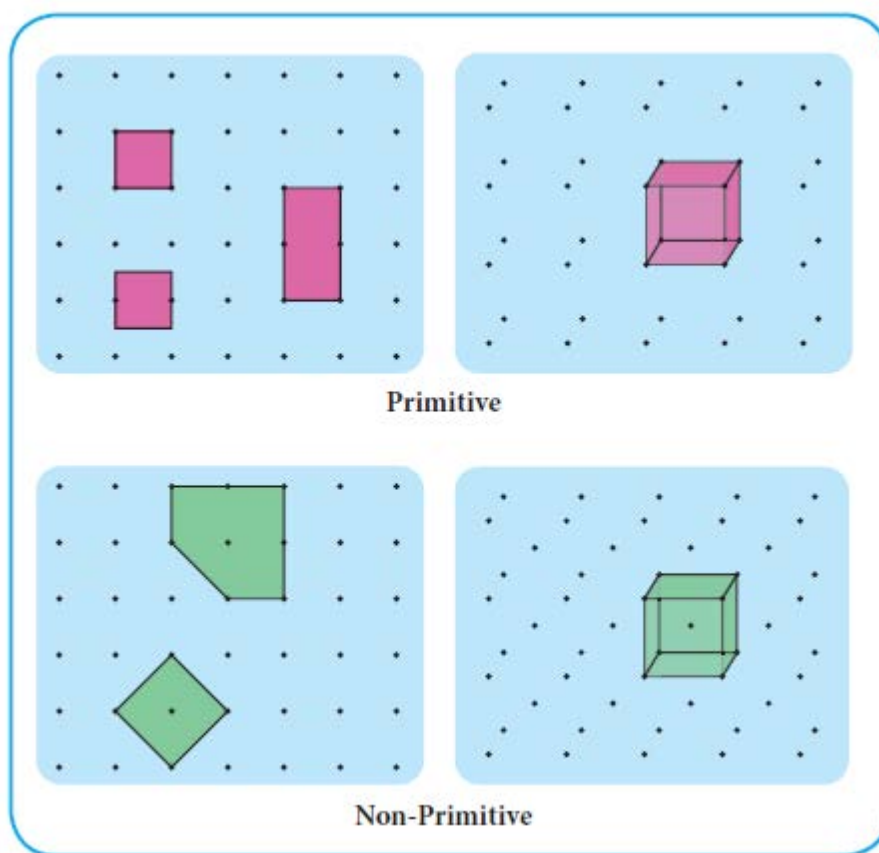


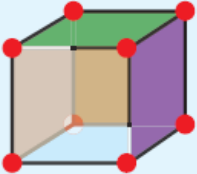
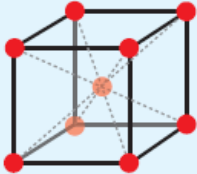
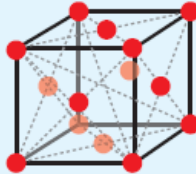
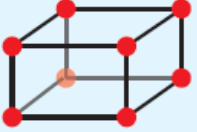
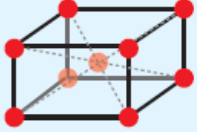
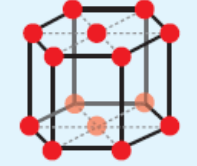
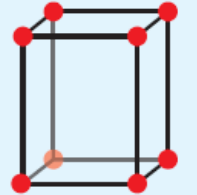
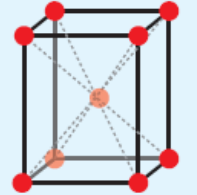
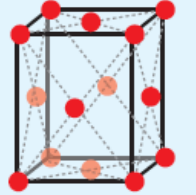
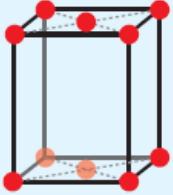
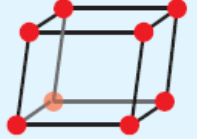
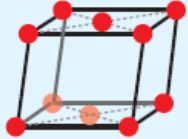
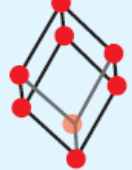
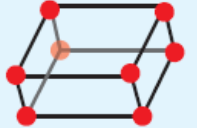
Primitive and non-primitive unit cell

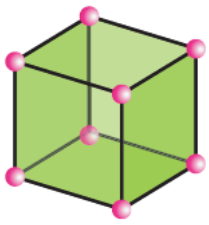
There are two types of unit cells: primitive and non-primitive. A unit cell that contains only one lattice point is called a primitive unit cell, which is made up from the lattice points at each of the corners.

In case of non-primitive unit cells, there are additional lattice points, either on a face of the unit cell or within the unit cell.

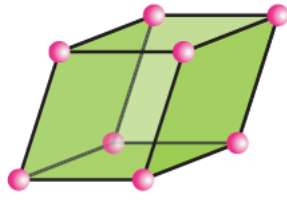


There are seven primitive crystal systems; cubic, tetragonal, orthorhombic, hexagonal, monoclinic, triclinic and rhombohedral. They differ in the arrangement of their crystallographic axes and angles. Corresponding to the above seven, Bravis defined 14 possible crystal systems as shown in the figure.

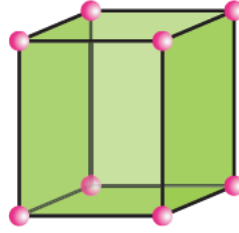
cubic	 <p>primitive</p>	 <p>body centered</p>	 <p>face centered</p>
tetragonal	 <p>primitive</p>	 <p>body centered</p>	
hexagonal			
orthorhombic	 <p>primitive</p>	 <p>body centered</p>	 <p>face centered</p>  <p>basis face centered</p>
monoclinic	 <p>primitive</p>		 <p>basis face centered</p>
trigonal			
triclinic			



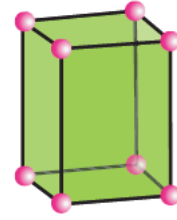
Cubic
 $a = b = c$
 $\alpha = \beta = \gamma = 90^\circ$



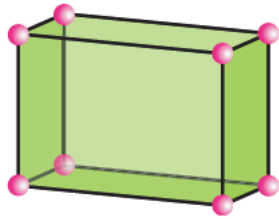
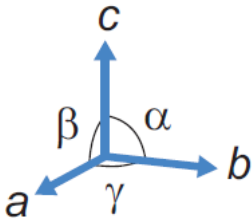
Rhombohedral
 $a = b = c$
 $\alpha = \beta = \gamma \neq 90^\circ$



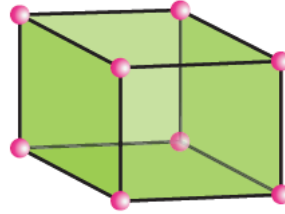
Hexagonal
 $a = b \neq c$
 $\alpha = \beta = 90^\circ, \gamma = 120^\circ$



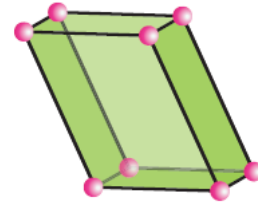
Tetragonal
 $a = b \neq c$
 $\alpha = \beta = \gamma = 90^\circ$



Orthorhombic
 $a \neq b \neq c$
 $\alpha = \beta = \gamma = 90^\circ$



Monoclinic
 $a \neq b \neq c$
 $\alpha = \gamma = 90^\circ, \beta \neq 90^\circ$



Triclinic
 $a \neq b \neq c$
 $\alpha \neq \beta \neq \gamma \neq 90^\circ$